

TOTAL SYNTHESIS OF THE DISACCHARIDE OF BLEOMYCIN,

2-O-(α -D-MANNOPYRANOSYL)-L-GULOPIRANOSE

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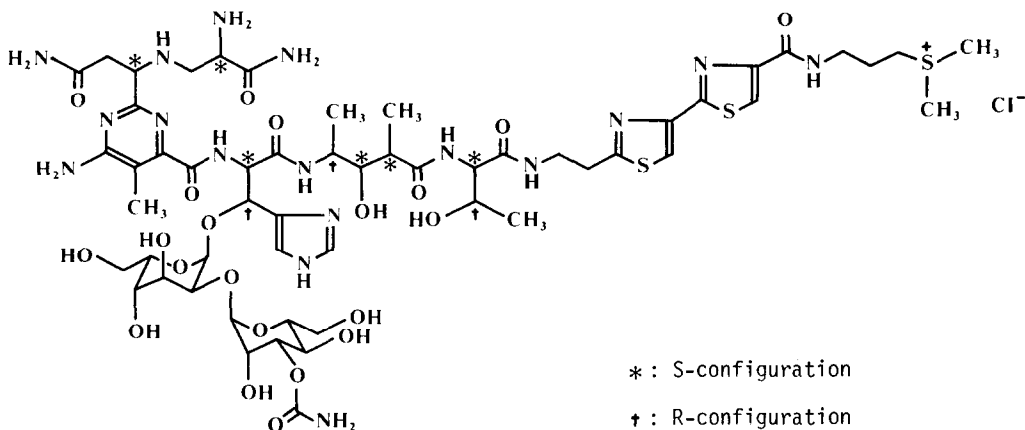
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Summary: 2-O-(α -D-Mannopyranosyl)-L-gulopyranose, the sugar portion of bleomycin has been synthesized.

Bleomycin (BLM) is an antitumor antibiotic clinically used in the treatment of several types of cancer.¹ The definitive structure has been established^{2,3} as shown below.



Total synthesis of BLM attracted attention and the peptide part of BLM has recently been synthesized by Takita et al.⁴ This paper describes the first synthesis of the disaccharide part of BLM, namely 2-O-(α -D-mannopyranosyl)-L-gulose⁵ (9).

As for a synthetic block for total synthesis of BLM, L-gulose⁶ and its 3,4-di-O-benzyl derivatives⁷ could be the candidates. However, these compounds seem not suitable because the

introduction of the mannose portion into their 2-hydroxyl groups is extremely difficult. We chose, instead, to prepare 2-O-(\underline{D} -mannosyl)- \underline{L} -gulose (9) as a suitable precursor for total synthesis of BLM.

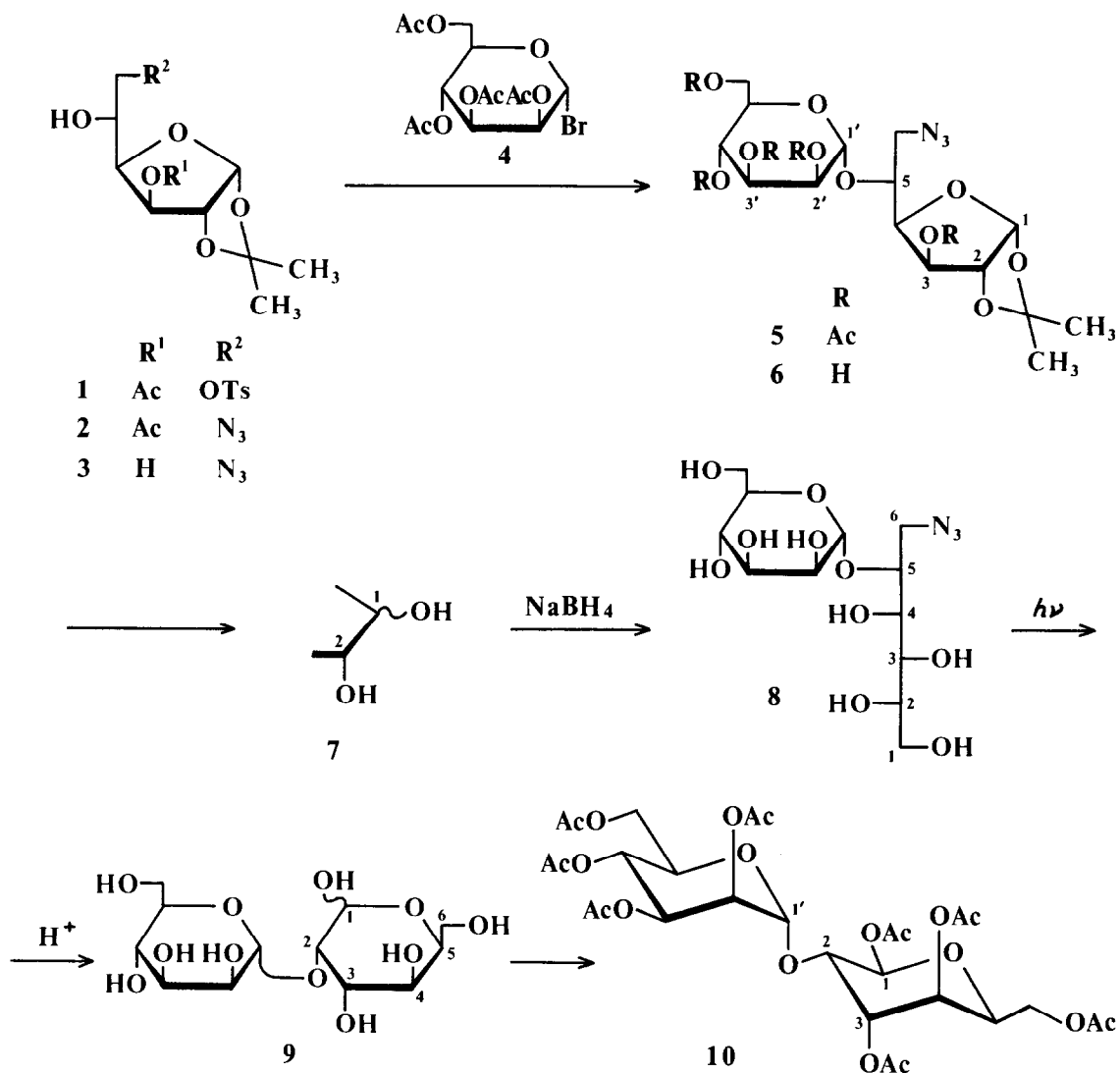
The characteristic points in the disaccharide synthesis are the early-stage condensation of \underline{D} -mannose to the 5-OH group of a \underline{D} -glucofuranose derivative, and successive head-to-tail inversion of the \underline{D} -glucofuranose moiety, the C-5 of the \underline{D} -glucose moiety being converted to C-2 in the new-born \underline{L} -gulose moiety as a result of the inversion, thus completing the synthesis of 9.

Readily preparable 3-O-acetyl-1,2-O-isopropylidene-6-O-tosyl- \underline{D} -glucofuranose⁸ (1) was treated with sodium azide in N,N-dimethylformamide to give the corresponding 6-azido derivative (2) (syrup, 71%), $[\alpha]_D^{25} - 6^\circ$ (c 1, chloroform), i.r. 2100 cm^{-1} (N_3). This compound is apt to suffer 3-O- to 5-O-acetyl migration to give 5-O-acetyl isomer (2'), during the reaction. However, condensation of 2 with 2,3,4,6-tetra-O-acetyl- α - \underline{D} -mannopyranosyl bromide⁹ (4) in CH_2Cl_2 in the presence of $\text{Hg}(\text{CN})_2$ gave 3-O-acetyl-6-azido-6-deoxy-1,2-O-isopropylidene-5-O-(2,3,4,6-tetra-O-acetyl- α - \underline{D} -mannopyranosyl)- \underline{D} -glucofuranose (5) (52%, after chromatography), $[\alpha]_D^{25} - 36^\circ$ (c 1, chloroform); Found (Calcd): C, 48.80 (48.62), H, 5.66 (5.71), N, 6.48 (6.81). The possibility of condensation between 2' and 4 was negligible, because 5 retains, in its $^1\text{H-NMR}$ spectrum (in CDCl_3), almost identical δ and J values with those of 2 in respect to $(\text{CH}_3)_2\text{C}$, (δ 1.34, 1.58), 3-O-Ac (2.17), H-1 (5.90 d, $J_{1,2}$ 3.5 Hz), H-2 (4.56 d) and H-3 (5.32 d, $J_{3,4}$ 3.0 Hz). Deacetylation of 5 gave 6 (85%) as needles, m.p. $170 - 170.5^\circ$, $[\alpha]_D^{25} + 34^\circ$ (c 1, water); $^1\text{H-NMR}$ (in D_2O): δ 5.09 (a slightly unresolved s, H-1'), 6.00 (d, $J_{1,2}$ 3.5 Hz, H-1). Carbon-13 NMR spectrum of 6 was shown in Table 1. The assignments were made on the basis of comparison with the shift data of 6-azido-6-deoxy-1,2-O-isopropylidene- \underline{D} -glucofuranose (3) (deacetylated product of 2) [needles, m.p. $107.5 - 108.5^\circ$, $[\alpha]_D^{25} - 11^\circ$ (c 1, water)] and of methyl α - \underline{D} -mannopyranoside (Me α - \underline{D} -Man). Down-field shift¹⁰ (7.4 ppm) of C-5 from that position in 3 shows that glycosidation occurred at the carbon. The anomeric configuration (α - \underline{D}) of the mannoside was verified by the $J_{\text{C}1', \text{H}1'}$ value (that of β - \underline{D} -mannoside is expected¹¹ to be ca. 160 Hz) and the shift-values (C-1' - C-6'); methyl β - \underline{D} -mannoside gave different values.^{12,13}

Acidic hydrolysis of 6 gave deacetonated product (7), which was reduced with sodium borohydride to give a non-reducing sugar, 6-azido-6-deoxy-5-O-(α - \underline{D} -mannopyranosyl)- \underline{D} -glucitol (8) (69% from 6 after purification), $[\alpha]_D^{24} + 24^\circ$ (c 0.9, water); i.r. 2100 cm^{-1} ; $^1\text{H-NMR}$ (in D_2O): δ 5.07 (1 H d, $J_{1,2}$ 1.5 Hz, H-1').

Conversion of the $-\text{CH}_2\text{N}_3$ group of 8 to aldehyde group was successfully carried out by photolysis, which was first applied to carbohydrates by Horton et al.¹⁵ A 2.5% aqueous solution of 8 was, after bubbling nitrogen, irradiated (3000Å , 5.5 h) at room temperature, and, after addition of Dowex 50W x8 resin (H^+ form), the mixture was stirred for 1 h. During the reaction, intermediary imine¹⁵ was converted to an aldehyde and the latter condensed with the hydroxyl group at C-5 (former C-2 of glucofuranose) to form \underline{L} -gulopyranose. The crude product obtained after evaporation was purified by silica gel column-chromatography (Wakogel C-200, 2:1:1 butanol-acetic acid-water) to give 9 as a hygroscopic solid (48%) $[\alpha]_D^{25} + 96^\circ$ (c 1, water) (final value); Found (Calcd as hemihydrate): C, 41.01 (41.02), H, 6.46 (6.59).

Acetylation of **9** with acetic anhydride in pyridine gave 1,3,4,6-tetra-*O*-acetyl-2-*O*-(2,3,4,6-tetra-*O*-acetyl- α -D-mannopyranosyl)- β -L-gulopyranose (**10**) (82%), $[\alpha]_D^{25} + 36^\circ$ (c 0.5, chloroform); Found (Calcd); C, 49.30 (49.56), H, 5.69 (5.64). The $^1\text{H-NMR}$ data (Table 2) indicates the conformation as shown in **10**. The i.r. and the $^1\text{H-NMR}$ spectra of the compound were identical with those of the compound obtained by acetylation of natural decarbamoyl disaccharide.⁵ We are now challenging the total synthesis of bleomycin by utilizing the synthesized disaccharide (**9**).



References

1. H. Umezawa, *Progr. Biochem. Pharmacol.*, **11**, 18-27 (1976).
2. T. Takita, Y. Muraoka, T. Nakatani, A. Fujii, Y. Umezawa, H. Naganawa and H. Umezawa, *J. Antibiot.*, **31** 801-804 (1978).
3. T. Takita, Y. Muraoka, T. Nakatani, A. Fujii, Y. Iitaka and H. Umezawa, *J. Antibiot.*, **31**, 1073-1077 (1978).

Table 1. ^{13}C Chemical shift data^a
(62.9 MHz in D_2O) of $\underline{6}$

| | $\underline{6}$ | $\underline{3}^b$ |
|---------------------------------------|--------------------------------|--|
| C-1 | 105.5 | 105.4 |
| | ($^1J_{\text{C,H}}$ 186.0 Hz) | ($^1J_{\text{C,H}}$ 185.9 Hz) |
| -2 | 85.4 | 85.2 |
| -3 | 74.4 ^C | 74.1 |
| -4 | 79.5 | 80.9 |
| -5 | 75.3 | 67.9 |
| -6 | 53.4 | 55.0 |
| CH_3 | 26.0, 26.5 | 26.0, 26.4 |
| $\underline{\text{C}}(\text{CH}_3)_2$ | 113.6 | 113.5 |
| | | Me α - $\underline{\text{D}}$ -Man ^d |
| C-1' | 102.1 | 101.6 |
| | ($^1J_{\text{C,H}}$ 171.3 Hz) | ($^1J_{\text{C,H}}$ 170.7 Hz) |
| -2' | 71.1 | 70.7 |
| -3' | 71.1 | 71.4 |
| -4' | 67.4 | 67.5 |
| -5' | 74.0 ^C | 73.3 |
| -6' | 61.8 | 61.7 |
| OCH_3 | | 55.5 |

a: Ppm downfield from TMS calculated as $\delta^{\text{TMS}} = \delta^{\text{dioxane}} + 67.4$. b: Shift assignments were made based on the data of 1,2-O-isopropylidene- $\underline{\text{D}}$ -glucofuranose.¹⁴ c: The values of C-3 and C-5' may be interchangeable. d: Measured in D_2O ; the shift-values were almost identical to those reported.^{13,14}

Table 2. ^1H -NMR spectrum of $\underline{10}$
(250 MHz in CDCl_3)

| | | | |
|-----|---------|-------------|------|
| H-1 | 5.90 d | $J_{1,2}$ | 8.5 |
| -2 | 4.00 dd | $J_{2,3}$ | 3.5 |
| -3 | 5.45 t | $J_{3,4}$ | 3.5 |
| -4 | 5.02 dd | $J_{4,5}$ | 1.5 |
| -1' | 4.99 d | $J_{1',2'}$ | 1.8 |
| -2' | 5.10 dd | $J_{2',3'}$ | 3.4 |
| -3' | 5.16 dd | $J_{3',4'}$ | 10.1 |
| -4' | 5.28 t | | |

a: Assignments were made by decoupling technique as well as inspection of the signal patterns.

4. T. Takita, Y. Umezawa, S. Saito, H. Morishima, H. Umezawa, Y. Muraoka, M. Suzuki, M. Otsuka, S. Kobayashi and M. Ohno, *Tetrahedron Letters*, in press; Y. Umezawa, H. Morishima, S. Saito, T. Takita, H. Umezawa, S. Kobayashi, M. Otsuka, M. Narita and M. Ohno, *J. Am. Chem. Soc.*, 102, 6630-6631 (1980).
5. S. Omoto, T. Takita, K. Maeda, H. Umezawa and S. Umezawa, *J. Antibiot.*, 25, 752-754 (1972).
6. R. L. Whistler and J. N. BeMiller, *Methods Carbohydr. Chem.* Vol 1, p 137-139, Academic Press (1962); M. E. Evans and F. W. Parrish, *Carbohydr. Res.*, 28, 359-364 (1973); related references are cited therein.
7. D. K. Minster and S. M. Hecht, *J. Org. Chem.*, 43, 3987-3988 (1978).
8. M. Nakajima and S. Takahashi, *Agr. Biol. Chem.*, 31, 1079-1081 (1967).
9. F. Micheel and H. Micheel, *Chem. Ber.*, 63, 386-393 (1930).
10. T. Usui, N. Yamaoka, K. Matsuda, K. Tuzimura, H. Sugiyama and S. Seto, *J. Chem. Soc. Parkin I*, 1973, 2425-2432.
11. K. Bock, I. Lundt and C. Pedersen, *Tetrahedron Letters*, 1973, 1037-1040; K. Bock and C. Pedersen, *J. Chem. Soc. Parkin II*, 1974, 293-297.
12. A. S. Perlin, B. Casu and H. J. Koch, *Can. J. Chem.*, 48, 2596-2606 (1970).
13. T. E. Walker, R. E. London, T. W. Whaley, R. Barker and N. A. Matwiyoff, *J. Am. Chem. Soc.*, 98, 5807-5813 (1976).
14. D. W. Vyas, H. C. Jarrell and W. A. Szarek, *Can. J. Chem.*, 53, 2748-2754 (1975).
15. D. Horton, A. E. Luetzow and J. C. Wease, *Carbohydr. Res.*, 8, 366-367 (1968).

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